

# True Tolerance Factor Effects in $Ln_{1.85}^{3+}M_{0.15}^{2+}CuO_4$ Superconductors

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A series of seven  $Ln_{1.85}^{3+}M_{0.15}^{2+}CuO_4$  samples has been prepared in which mixtures of  $Ln = La$  and  $Nd$  and  $M = Ca, Sr,$  and  $Ba$  are used to increase the average radius from 1.180 to 1.223 Å (equivalent to an increase of tolerance factor from 0.856 to 0.871) while the size variance is fixed at a constant value of  $\sigma^2 = 0.0020 \text{ \AA}^2$ . X-ray powder diffraction is used to study structural changes, and ac magnetic susceptibility, electrical resistivity, and thermoelectric power measurements show the variation of physical properties as a true function of the tolerance factor. © 2000 Academic Press

## INTRODUCTION

The superconducting  $A_2CuO_4$  cuprates are intergrowths of  $CuO_2$  copper oxide planes and insulating double  $AO$  layers ( $A$  is a mixture of  $Ln^{3+}$  rare-earth and  $M^{2+}$  alkaline earth cations). For the  $La_2CuO_4$  type structure to be stable there should be sufficient matching between the bond lengths in each layer. This is traditionally measured by the perovskite tolerance factor,  $t = (\langle r_A \rangle + r_o) / \sqrt{2}(r_{Cu} + r_o)$  where  $\langle r_A \rangle$  is the mean  $A$  cation radius and  $r_{Cu}$  and  $r_o$  are the copper and oxide ion radii. The mismatch between  $La-O$  and  $Cu-O$  bonds is large so that  $t < 1$ , placing the  $Cu-O$  bonds under compression and  $La-O$  under tension. This can be alleviated through tilting of the  $CuO_6$  octahedra or by substitution with a larger cation. Doping with an alkaline earth-metal cation has the added effect of removing an antibonding electron from the  $Cu-O$  bond. If  $La$  is replaced by a smaller ion the tolerance factor is reduced further and eventually a transition to the  $T'$  ( $Nd_2CuO_4$  type) structure occurs (1). The physical behavior of the single-layer high  $T_c$  cuprates is also strongly dependent on the properties of the insulating layer between the copper oxide planes (2). By altering the  $A$  site composition in T-type ( $Ln_{1-x}^{3+}M_x^{2+}$ ) $_2CuO_4$  the superconducting properties, as well as structural transitions, can be tuned (3, 4).

Recently it has been shown that the cation size variance  $\sigma^2 (= \langle r_A^2 \rangle - \langle r_A \rangle^2)$  is also important in controlling the

behavior of ( $Ln_{1-x}^{3+}M_x^{2+}$ ) $_2CuO_4$  cuprates (5). If the doping level is fixed (at the optimum  $x = 0.075$  value) and  $\langle r_A \rangle$  is held constant, then the superconducting critical temperature  $T_c$  decreases linearly with  $\sigma^2$  while the structural transition from the high temperature tetragonal (HTT,  $I4/mmm$  symmetry) to the low temperature orthorhombic (LTO1,  $Abma$  symmetry) structure increases linearly (6). Systematic changes in the carrier densities in the normal and superconducting states are also evidenced from magnetic and transport measurements (6, 7).

Previous studies of the  $A$  cation size (tolerance factor) effects in ( $Ln_{1-x}^{3+}M_x^{2+}$ ) $_2CuO_4$  superconductors did not allow for the size variance effect, and so we have undertaken this study of a series of optimally doped ( $x = 0.075$ ) materials in which  $\sigma^2$  is also fixed at a convenient value ( $0.002 \text{ \AA}^2$ ) in order to demonstrate the true tolerance factor dependence of the structural and superconducting properties.

## EXPERIMENTAL

A series of seven samples were prepared, each with  $\sigma^2 = 0.002 \text{ \AA}^2$  and covering a range of  $1.180 \leq \langle r_A \rangle \leq 1.223 \text{ \AA}$  (calculated using standard nine-coordinate ionic radii (8)). Sample compositions are given in Table 1. Appropriate amounts of  $La_2O_3$ ,  $Nd_2O_3$ ,  $CaCO_3$ ,  $SrCO_3$ ,  $BaCO_3$ , and  $CuO$  were ground together, pressed into pellets, and heated at  $950^\circ\text{C}$  for 24 h to achieve decarbonation. The samples were reground and heated at  $1050^\circ\text{C}$  for 200 h with several intermediate regrindings. After the final heating, the pellets were annealed in flowing oxygen at  $400^\circ\text{C}$  for 100 h.

The samples were characterized by powder X-ray diffraction scans, collected over 8 h on a STOE STADI P diffractometer from  $20 \leq 2\theta \leq 90^\circ$  using monochromated  $CuK\alpha_1$  radiation with a step size of  $0.01^\circ$ . Rietveld fits were performed using the GSAS package (9) to determine cell parameters and phase fractions (Tables 1 and 2).

Only samples 4–7 were found to contain a single  $La_2CuO_4$  type phase and these were further characterized by thermogravimetric analysis from 20 to  $800^\circ\text{C}$  under

**TABLE 1**  
**A Site Composition, Average Radius  $\langle r_A \rangle$ , Tolerance Factor ( $t$ ), and Phase Fractions from X-ray Diffraction Analysis for Seven  $A_2CuO_4$  Samples**

Sample no.	A site composition	$\langle r_A \rangle$ (Å)	$t$	Phase fractions (%)			
				T'	La <sub>2</sub> CaCu <sub>2</sub> O <sub>6</sub>	T (LTO1)	T (HTT)
1	La <sub>0.185</sub> Nd <sub>0.740</sub> Ca <sub>0.050</sub> Sr <sub>0.008</sub> Ba <sub>0.017</sub>	1.180	0.856	86.4(2)	13.6(2)	0	0
2	La <sub>0.369</sub> Nd <sub>0.556</sub> Ca <sub>0.045</sub> Sr <sub>0.016</sub> Ba <sub>0.014</sub>	1.190	0.860	60.0(1)	15.0(1)	25.0(1)	0
3	La <sub>0.490</sub> Nd <sub>0.435</sub> Ca <sub>0.010</sub> Sr <sub>0.057</sub> Ba <sub>0.008</sub>	1.200	0.863	14.7(4)	0	85.3(4)	0
4	La <sub>0.640</sub> Nd <sub>0.285</sub> Ca <sub>0.035</sub> Sr <sub>0.024</sub> Ba <sub>0.016</sub>	1.206	0.865	0	0	100	0
5	La <sub>0.704</sub> Nd <sub>0.221</sub> Ca <sub>0.013</sub> Sr <sub>0.047</sub> Ba <sub>0.015</sub>	1.212	0.867	0	0	100	0
6	La <sub>0.797</sub> Nd <sub>0.128</sub> Ca <sub>0.010</sub> Sr <sub>0.046</sub> Ba <sub>0.019</sub>	1.218	0.869	0	0	100	0
7	La <sub>0.925</sub> Ca <sub>0.037</sub> Sr <sub>0.008</sub> Ba <sub>0.030</sub>	1.223	0.871	0	0	0	100

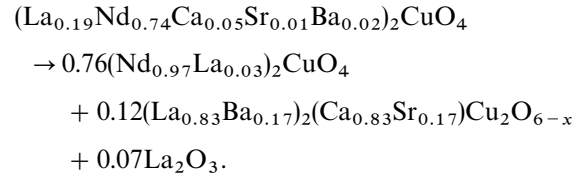
H<sub>2</sub>/N<sub>2</sub>, and the following physical measurements. AC susceptibilities were recorded for powdered samples in a field of 1 G, with frequency 333 Hz, in the temperature range 4.2–50 K to determine  $T_c$  and the diamagnetic fractions. Four probe resistivity measurements were made on sintered bars in the temperature range 8–300 K, with a measuring current of 100  $\mu$ A. The room-temperature thermoelectric power  $S(290)$  was determined using the apparatus developed by Obertelli *et al.* (10) and the corresponding hole concentration was estimated from standard data (11). Results of these measurements are shown in Table 3.

## RESULTS AND DISCUSSION

### a. Structure Analysis

The major phases found from the powder X-ray diffraction analyses of the seven samples are shown in Table 1, although small proportions (<5%) of additional phases may not be observed. The analyses reveal changes in the  $A_2CuO_4$  structure type and the phase assemblage as  $\langle r_A \rangle$  increases from 1.18 to 1.223 Å. The structure type changes from T' (Nd<sub>2</sub>CuO<sub>4</sub> type) to T (La<sub>2</sub>CuO<sub>4</sub> type), as is well known from previous studies, e.g., Ref. 1. In the T' structure, the CuO<sub>2</sub> planes are under tension and so can be electron

doped giving the  $n$ -type superconductors such as Nd<sub>2-x</sub>Ce<sub>x</sub>CuO<sub>4</sub> (12,13). However, they are not amenable to  $p$ -type doping, so that in samples 1 and 2 the  $M^{2+}$  cations segregate into a secondary La<sub>2</sub>CaCu<sub>2</sub>O<sub>6</sub>-type phase (14, 15), which can accommodate Sr on Ca sites and Ba on the La sites. Assuming the latter cation distribution allows the phase segregation in sample 1 to be written as



The expected Nd<sub>2</sub>CuO<sub>4</sub>-type/La<sub>2</sub>CaCu<sub>2</sub>O<sub>6</sub>-type ratio of 0.76/0.12 = 6.33 compares extremely well with the observed ratio (Table 1) of 86.4/13.6 = 6.35, although the additional La<sub>2</sub>O<sub>3</sub> is not observed and this may have formed a poorly crystalline carbonate or hydroxide. The presence of the secondary La<sub>2</sub>CaCu<sub>2</sub>O<sub>6</sub>-type phase shows that  $\sigma^2$  is not a good functional for the  $\langle r_A \rangle < 1.20$  Å samples in this study, as the observed phases depend upon the proportions of the A-type cations, and not just the statistics of the overall A cation distribution.

The change from T' to T-type structure is seen in samples 2 and 3 which contain both phases and have tolerance factors  $0.860 \leq t \leq 0.863$ . A study by Manthiram and Goodenough (1) of the tolerance factor effects on the structures of undoped La<sub>2-y</sub>Nd<sub>y</sub>CuO<sub>4</sub> solid solutions found the T' to T transition at a slightly lower tolerance factor ( $t \leq 0.859$ ), although in this series there is also a varying  $\sigma^2$  of magnitude  $0.0007 \cdot y \cdot (2 - y)$  Å<sup>2</sup>.

The T phases in samples 2–6 were found to have the LTO1 structure, but 7 has the undistorted HTT arrangement. Rietveld fits were used to refine the lattice parameters (Table 2) and a typical profile plot is shown in Fig. 1. As  $\langle r_A \rangle$  increases, the cell volume rises and the orthorhombicity  $(a - b)/(a + b)$  decreases smoothly (Fig. 2). The Cu–O bonds are under compression and the A–O bonds are under

**TABLE 2**  
**Structure Type, Lattice Parameters, Cell Volume ( $\times 0.5$  for LTO1 Phases), and Orthorhombicity ( $o$ ) for the Majority  $A_2CuO_4$  Phases in Samples 1–7**

Sample no.	Structure type	$a$ (Å)	$b$ (Å)	$c$ (Å)	$V$ (Å <sup>3</sup> )	$o$
1	T'	3.95826(9)	—	12.2481(3)	191.90(1)	0
2	T'	3.96271(5)	—	12.2822(3)	192.87(1)	0
3	T(LTO1)	5.3659(1)	5.3129(1)	13.0686(2)	186.29(1)	0.00992
4	T(LTO1)	5.3530(2)	5.3120(2)	13.0772(6)	185.93(2)	0.00768
5	T(LTO1)	5.3610(2)	5.3314(2)	13.1482(6)	187.90(2)	0.00554
6	T(LTO1)	5.3374(2)	5.3232(2)	13.1609(5)	186.97(1)	0.00267
7	T(HTT)	3.7857(1)	—	13.2272(2)	189.56(1)	0

**TABLE 3**  
**Results of Physical Measurements on  $A_2\text{CuO}_4$  Samples 4–7:  $T_c$  (by ac susceptibility and resistivity measurements),  $F$ ,  $\rho_{\text{res}}$ ,  $S(290)$ , and Hole Concentration,  $p$**

Sample no.	$\langle r_A \rangle$ (Å)	Oxygen content	$T_c$ (K) (ACS)	$F$ (%) at 5 K	$T_c$ (K) (resistivity)	$\rho_{\text{res}}$ (mΩ cm)	$S(290)$ ( $\mu\text{V K}^{-1}$ )	$p$
4	1.206	3.97(5)	—	—	6.7	0.55	28.7(5)	0.145
5	1.212	3.96(8)	—	—	8.4	0.52	26.2(5)	0.152
6	1.218	3.94(6)	21	4.7	18.1	0.27	25.0(5)	0.154
7	1.223	4.05(8)	20	6.2	19.0	0.15	44.6(5)	0.124

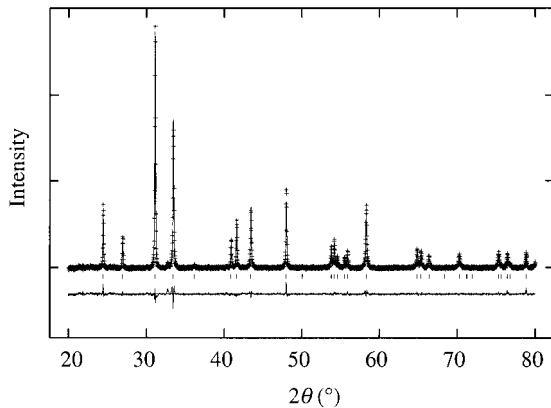
tension in the T structure, so that increasing  $\langle r_A \rangle$  reduces both of these strains, decreasing the tilting of the  $\text{CuO}_6$  octahedra and the cell orthorhombicity, ultimately resulting in a transition to the tetragonal structure for  $\langle r_A \rangle \approx 1.223$  Å ( $t = 0.87$ ).

### b. Physical Properties

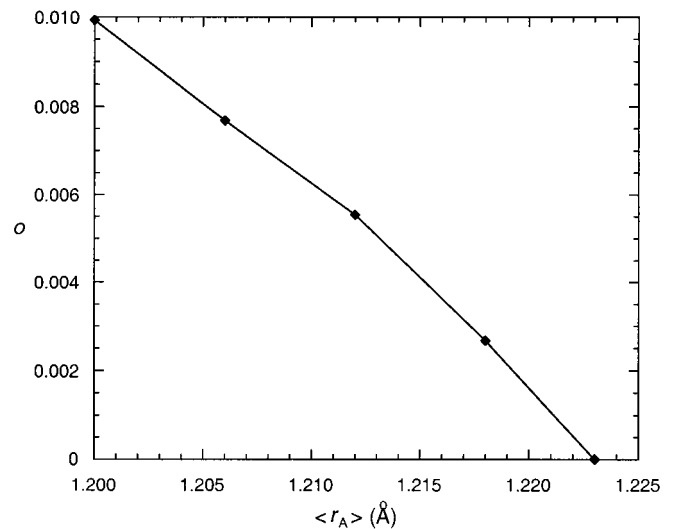
The physical properties of the single T phase samples 4–7 were investigated. Thermogravimetric analyses show that all four samples are oxygen stoichiometric within error, so the nominal hole doping level is fixed by the A cation compositions at 0.15 holes per Cu atom. Figure 3 shows the AC susceptibility curves for these samples. Samples 6 and 7 exhibit clear superconducting transitions with appreciable diamagnetic fractions at 5 K characteristic of bulk superconductivity, sample 4 shows a very small downturn around 10 K (see inset), and sample 5 shows no evidence of a diamagnetic transition. Four-probe electrical resistivity data (Fig. 4) show transitions for all four samples, suggesting that there are superconducting pathways percolating through samples 4 and 5, although bulk superconductivity is not observed by magnetic measurements. This is apparently at odds with previous results for a sample with similar  $\langle r_A \rangle$

( $= 1.212$  Å) and  $\sigma^2$  ( $= 0.019$  Å<sup>2</sup>) to sample 5 (5, 6). This sample had a magnetic  $T_c$  of 23 K and a fractional diamagnetism of  $F = 4\%$  at 5 K, after annealing under 550 atm oxygen pressure, whereas the present samples were treated only under 1 atm oxygen pressure. It is clear that high-pressure annealing improves the superconducting properties of  $(\text{Ln}_{1-x}^{3+}\text{M}_x^{2+})_2\text{CuO}_4$  materials, principally by making the samples less granular.

All of the samples 4–7 show resistive upturns on cooling toward  $T_c$ , characteristic of underdoped cuprates, and the effect becomes less pronounced as  $\langle r_A \rangle$  decreases. Values for the zero resistance  $T_c$  and the residual resistance extrapolated from the high-temperature linear regime to  $T = 0$ ,  $\rho_{\text{res}}$ , were taken from these measurements and are given in Table 3. The ambient temperature thermoelectric power  $S(290)$  is known to correlate with hole concentration  $p$  in systems such as  $\text{La}_{2-x}\text{Sr}_x\text{CuO}_4$  and this correlation was used to estimate the  $p$  values shown in Table 3 which give a good measure of relative hole doping although the absolute accuracy is not as good. The value for sample 7 is clearly anomalous; the reason for this is not clear although



**FIG. 1.** Observed, calculated, and difference 290 K X-ray diffraction profiles for  $(\text{La}_{0.925}\text{Ca}_{0.037}\text{Sr}_{0.008}\text{Ba}_{0.030})_2\text{CuO}_4$  (sample 7).



**FIG. 2.** Variation of orthorhombicity ( $o$ ) with  $\langle r_A \rangle$  for the T type phases in samples 3–7.

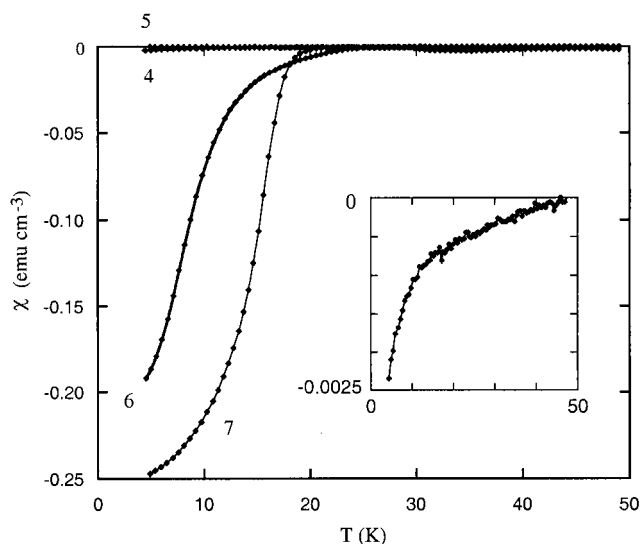


FIG. 3. Alternating current magnetic susceptibility data for samples 4-7. The inset shows the weak diamagnetism of sample 4.

we note that a sample of the same composition prepared in a previous study (of  $\sigma^2$  effects at a constant  $\langle r_A \rangle = 1.223 \text{ \AA}$  (6)) had a  $T_c$  of 25 K which is more in keeping with the  $T_c$  trend for samples 3-6.

Despite some anomalies in the physical properties of sample 7, trends in the data in Table 3 are apparent. As  $\langle r_A \rangle$  decreases at constant  $\sigma^2$ ,  $T_c$  falls,  $\rho_{\text{res}}$  increases, and (excepting sample 7)  $S(290)$  increases slightly. These changes are all consistent with a slight reduction in the free hole concentration in the normal state, the magnitude of which can be seen from the estimates of  $p$  in Table 3. A corresponding fall in the superfluid density below  $T_c$  is evidenced by the decrease

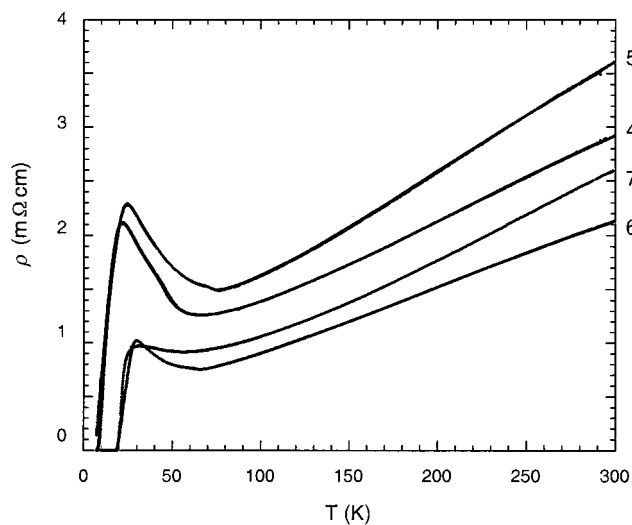


FIG. 4. Resistivity data for samples 4-7.

in the fractional diamagnetism of the samples (and hence an increase in the penetration depth).

The above trends in physical properties with decreasing  $\langle r_A \rangle$  at constant  $\sigma^2$  are the same as those previously observed in  $Ln_{1-x}^{3+}M_x^{2+}CuO_4$  materials with increasing  $\sigma^2$  at constant  $\langle r_A \rangle$  (6). The trends suggest that an increasing proportion of the current carrying holes is trapped as  $\langle r_A \rangle$  decreases or  $\sigma^2$  increases. This is in keeping with a simple empirical model for the effects of  $A$  cation size distributions proposed previously (5, 16), in which it is proposed that the temperatures of electronic transitions such as  $T_c$  in perovskite-type compounds decrease linearly with  $\sigma^2$  at constant  $\langle r_A \rangle$ , and with  $(r_A^0 - \langle r_A \rangle)^2$  at constant  $\sigma^2$  ( $r_A^0$  is the  $A$  cation radius that gives a tolerance factor of  $t = 1$ ). However, the range of  $\langle r_A \rangle$  over which the T type phase exists is too narrow to enable the latter quadratic function to be tested quantitatively here.

## CONCLUSIONS

This study shows that true effects of varying tolerance factor  $t$  in doped perovskites such as the  $(Ln_{1-x}^{3+}M_x^{2+})_2CuO_4$  superconductors can be measured by fixing the size variance  $\sigma^2$ . This is achieved using mixtures of several cations such as those in Table 1 to give the desired size distribution and average charge. This approximation works well provided a single homogenous phase is synthesized, but when a mixture of phases results, as for samples 1 and 2, then the assemblage is a function of the proportions of individual cations rather than the statistics of the total distribution.

The qualitative results are in agreement with many earlier studies of  $(Ln_{1-x}^{3+}M_x^{2+})_2CuO_4$  superconductors, e.g., Ref. (1), as increasing the Cu-O-Cu bond angles toward  $180^\circ$  is known to enhance superconductivity (17). The structure type changes from the T' to the T type with increasing  $t$ , at  $t = 0.86$  for the fixed values of  $x = 0.075$  and  $\sigma^2 = 0.002 \text{ \AA}^2$  used here. The available A cations only enable T-type structures to be prepared up to  $t = 0.87$ ; however, large changes in structure and property occur over the range  $t = 0.865-0.871$  studied for samples 4-7. The structure changes from orthorhombic to tetragonal, and the superconducting properties and normal state transport parameters all evidence a significant decrease in carrier trapping as  $t$  increases over this small range. These changes are the same as those previously observed when  $\sigma^2$  is increased at constant  $t$  (5), although the rates of change were found to depend strongly on the value of  $t$ , and it is probable that different fixed  $\sigma^2$  values will give different rates of change of property with  $t$  (all at constant  $x$ ). Ultimately, this method of parameterization can be used to construct a four-dimensional phase diagram using  $T$ ,  $x$ ,  $t$  or  $\langle r_A \rangle$ , and  $\sigma^2$  as variables to display the structure types and physical properties of all  $(Ln_{1-x}^{3+}M_x^{2+})_2CuO_4$  materials.

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## REFERENCES

1. A. Manthiram and J. B. Goodenough, *J. Solid State Chem.* **87**, 402 (1990).
2. J. B. Torrance, Y. Tokura, A. I. Nazzari, A. Beziinge, T. C. Huang, and S. S. P. Parkin, *Phys. Rev. Lett.* **61**, 1127 (1988).
3. D. M. Paul, G. Balakrishnan, N. R. Bernhoeft, W. I. F. David, and W. T. A. Harrison, *Phys. Rev. Lett.* **58**, 1979 (1987).
4. J. D. Axe, A. H. Moudden, D. Hohlwein, D. E. Cox, K. M. Mohanty, A. R. Moodenbaugh, and Y. W. Xu, *Phys. Rev. Lett.* **62**, 2751 (1989).
5. J. P. Attfield, A. L. Kharlanov, and J. A. McAllister, *Nature* **394**, 157 (1998).
6. J. A. McAllister and J. P. Attfield, *Phys. Rev. Lett.* **83**, 3289 (1999).
7. J. P. Attfield, A. L. Kharlanov, J. A. McAllister, and L. M. Rodriguez-Martinez, *Mater. Res. Soc. Symp. Proc.* **547**, 15 (1999).
8. R. D. Shannon, *Acta Crystallogr. A* **32**, 751 (1976).
9. A. C. Larson and R. B. Von Dreele, *Los Alamos National Laboratory Rep.* **LA-UR-86-748**, 1987.
10. S. D. Obertelli, J. R. Cooper, and J. L. Tallon, US Patent 5,619,141 (08/04/97).
11. S(290) v hole concentration data supplied by Dr. J. R. Cooper and C. Le Borne.
12. Y. Tokura, H. Takagi, and S. Uchida, *Nature* **337**, 345 (1989).
13. H. Sawa, S. Suzuki, M. Watanabe, J. Akimitsu, H. Matsubara, H. Watabe, S. Uchida, K. Kokusho, H. Asano, F. Izumi, and E. Takayama-Muromachi, *Nature* **337**, 347 (1989).
14. A. R. Moodenbaugh, R. L. Sabatini, Y. W. Xu, J. Ochab, and J. G. Huber, *Physica C* **198**, 103 (1992).
15. H. Deng, C. Don, H. Chen, F. Wu, S. L. Jia, J. C. Shen, and Z. X. Zhao, *Physica C* **313** 285 (1999).
16. J. P. Attfield, *Chem. Mater.* **10**, 3239 (1998).
17. B. Dabrowski, Z. Wang, K. Rogach, J. D. Jorgensen, R. L. Hitterman, J. L. Wagner, B. A. Hunter, P. G. Radaelli, and D. G. Hinks, *Phys. Rev. Lett.* **76**, 1348 (1996).